

Crystal structure of the two-dimensional framework $[\text{Mn}(\text{salen})]_{4n}[\text{Re}_6\text{Te}_8(\text{CN})_6]_n$ [salen = N,N' -ethylenebis(salicylideneaminato)]

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The first complex-bridged cyano-rhenium cluster framework material $[\text{Mn}(\text{salen})]_{4n}[\text{Re}_6\text{Te}_8(\text{CN})_6]_n$, with a two-dimensional layered structure is prepared by the reaction of a methanol solution of $[\text{Mn}(\text{salen})]\text{ClO}_4$ with an aqueous solution of $\text{Na}_4[\text{Re}_6\text{Te}_8(\text{CN})_6]$.

The design of extended structures based on molecular complexes has received much attention. Recently, there have been many studies on developing engineered-supramolecular networks in an effective manner and designing highly dimensional molecular systems with adjustable structures.^{1,2} One way to obtain an extended structure is to utilize a molecular precursor that consists of multi-binding sites such as hexacyanoiron(III). For example, in MnL^+ ($L = 3\text{-MeO-salen}$,³ 5-Cl-salen ,^{4,5} or acacen ⁶) linkage occurs *via* hexacyanoiron(III) to form two-dimensional layers. These cyano-bridged mixed-metal compounds with ferromagnetic and antiferromagnetic interactions are two-dimensional network structures. There have also been extensive studies on the face-capped octahedral rhenium cluster, $[\text{Re}_6\text{Te}_8(\text{CN})_6]^{4-}$, which is analogous to hexacyanoiron(III). Cluster-expanded materials such as $\text{Fe}_4[\text{Re}_6\text{Te}_8(\text{CN})_6]_3 \cdot 27\text{H}_2\text{O}$, $\text{Ga}_4[\text{Re}_6\text{Se}_8(\text{CN})_6]_3 \cdot 38\text{H}_2\text{O}$, $[\text{Cd}_2(\text{H}_2\text{O})_4][\text{Re}_6\text{S}_8(\text{CN})_6] \cdot 14\text{H}_2\text{O}$, and $\text{Cs}_2[\text{trans-Fe}(\text{H}_2\text{O})_2]_3[\text{Re}_6\text{Se}_8(\text{CN})_6]_2 \cdot 18\text{H}_2\text{O}$, where simple metal ions link the Re clusters to form bigger cavities, resemble the Prussian Blue $\text{Fe}_4[\text{Fe}(\text{CN})_6]_3 \cdot 14\text{H}_2\text{O}$.⁷⁻⁹ These three-dimensional frameworks include plenty of water molecules as well as metal ions within their cavities. $[\text{Re}_6\text{S}_8(\text{CN})_6]^{4-}$ also coordinates to $\text{Mn}(\text{II})$ centers to form a neutral three-dimensional framework with isopropyl alcohol molecules.¹⁰ In order to develop new framework materials with a potential to create new spatial or chemical functions, we designed a new extended solid system based on the Re cluster. For our linking material, we chose the $[\text{Mn}(\text{salen})]^+$ complex instead of a simple metal ion since $[\text{Mn}(\text{salen})]^+$ is paramagnetic with a high-spin d^4 electronic configuration of $\text{Mn}(\text{III})$. Furthermore, it also has axial sites to which the cyano-rhenium clusters can link up. We report here the preparation of the first cyano-rhenium cluster which is linked to an Mn complex to provide an extended framework with a unique layered structure.

The face-capped octahedral rhenium cluster $\text{Na}_4[\text{Re}_6\text{Te}_8(\text{CN})_6]$ was prepared by a previous procedure.¹¹ The orange-red solid $\text{Na}_4[\text{Re}_6\text{Te}_8(\text{CN})_6]$ was purified by recrystallization using methanol-diethyl ether. After an aqueous solution of $\text{Na}_4[\text{Re}_6\text{Te}_8(\text{CN})_6]$ was carefully layered with a methanol solution of $[\text{Mn}(\text{salen})]\text{ClO}_4$, black crystals of $[\text{Mn}(\text{salen})]_{4n}[\text{Re}_6\text{Te}_8(\text{CN})_6]_n$ **1** formed directly at the interface. A quantitative yield¹² of **1** was obtained by mixing an aqueous solution of $\text{Na}_4[\text{Re}_6\text{Te}_8(\text{CN})_6]$ with a methanol solution of $[\text{Mn}(\text{salen})]\text{ClO}_4$. Compound **1** was not soluble in common solvents such as methanol, ethanol, water, diethyl ether, acetone, acetonitrile, or THF. The structure of **1** was determined by X-ray crystallography.¹³

The asymmetric unit contains half of an Re cluster, a bridged $[\text{Mn}(\text{salen})]^+$ unit, and a dangling $[\text{Mn}(\text{salen})]^+$ unit. The complete two-dimensional layered structure is generated by

the symmetry operations, $(-x + 3/2, y - 1/2, -z + 1/2)$, $(-x + 2, -y + 2, -z + 1)$, and $(-x + 3/2, y + 1/2, -z + 1/2)$. Four cyano groups of the Re cluster with C–N distances of 1.17(2)–1.18(2) Å are linked to $[\text{Mn}(\text{salen})]^+$ units, forming a two-dimensional layer. The remaining two cyano groups with C–N distance of 1.13(2) Å are bound to a $[\text{Mn}(\text{salen})]^+$ unit dangling outside the layer. The IR spectrum (KBr) of this compound showed two CN stretching peaks at 2073 and 2090 cm^{-1} , whereas the starting Re cluster had only one CN stretching peak at 2081 cm^{-1} . This result is consistent with the X-ray crystallographic data which shows two different types of cyano groups. Fig. 1(a) shows the rhenium cluster $[\text{Re}_6\text{Te}_8(\text{CN})_6]^{4-}$, surrounded by six $[\text{Mn}(\text{salen})]^+$ units. The Re–Re and Re–Te bonds range from 2.6842(7) to 2.7004(7) Å and from 2.685(1) to 2.703(1) Å, respectively. These are not significantly different from the starting rhenium cluster [The mean values of the Re–Re and Re–Te distances of the cyano-rhenium cluster are 2.681(3) and 2.694(2) Å, respectively]. The Mn(bridging

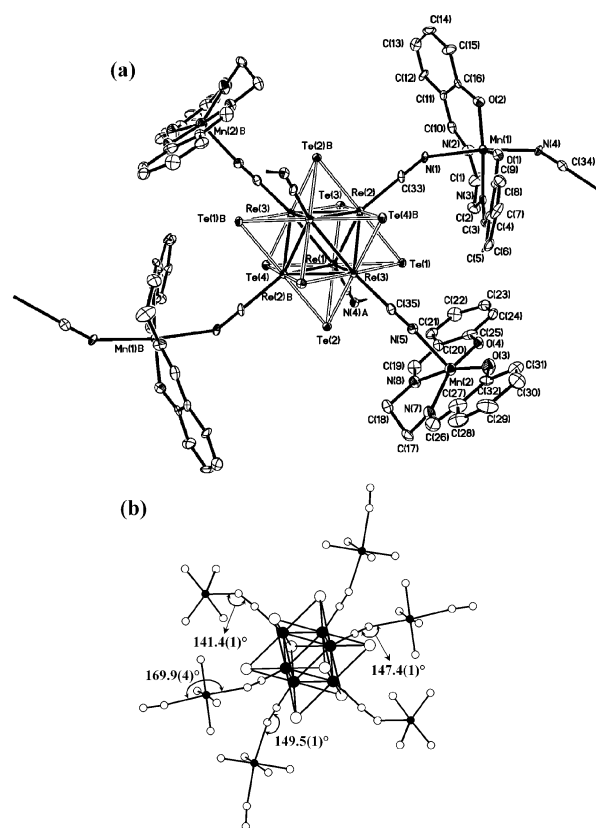


Fig. 1 (a) ORTEP drawing of the rhenium cluster $[\text{Re}_6\text{Te}_8(\text{CN})_6]^{4-}$ surrounded by six $[\text{Mn}(\text{salen})]^+$ units. Among the six $[\text{Mn}(\text{salen})]^+$ units bound to six rhenium atoms, only four are shown. (b) A simplified drawing with some bond angles around the rhenium cluster. All carbon atoms of the salen molecules have been omitted for clarity. Symmetry operations: A $(-x + 3/2, y - 1/2, -z + 1/2)$, B $(-x + 2, -y + 2, -z + 1)$.

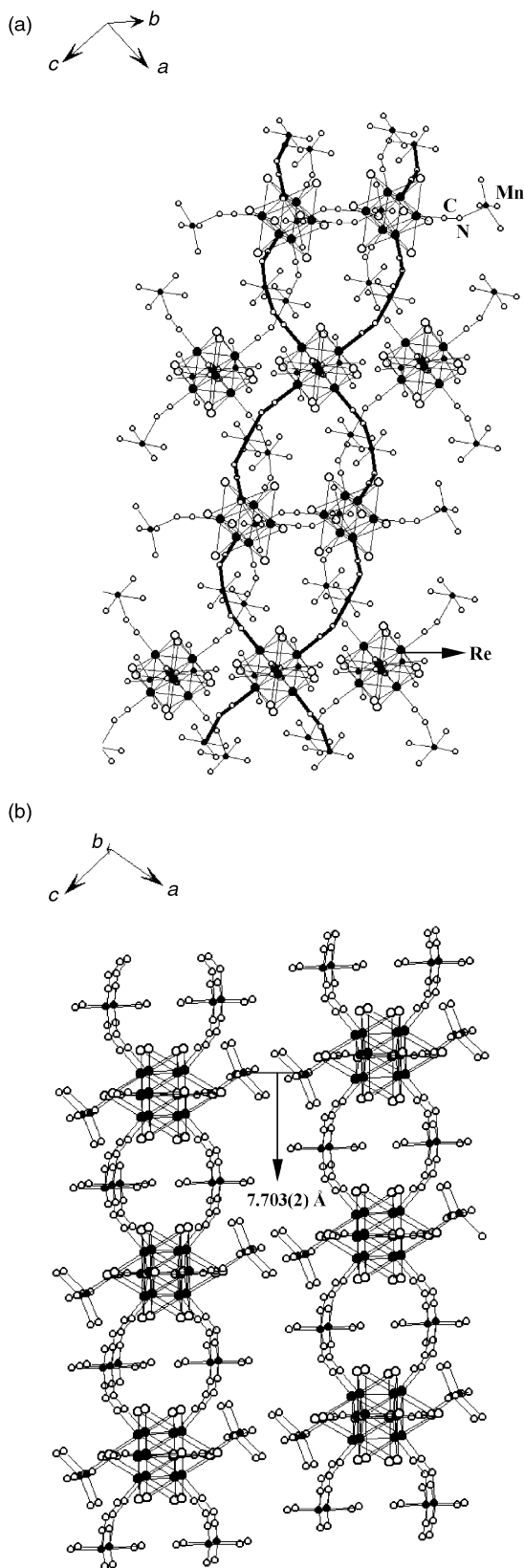


Fig. 2 (a) A two-dimensional layer containing $[\text{Mn}(\text{salen})]^+$ units linked by cyano-rhenium clusters. The twisted ropes interconnected by Re_6Te_8 clusters are shown. (b) Side view of two layers. All carbon atoms of the salen molecules are omitted for clarity.

$\text{Mn}(\text{unit})-\text{N}(\text{cyano group})$ distances are 2.27(1) and 2.28(1) Å, while the $\text{Mn}(\text{dangling Mn unit})-\text{N}(\text{cyano group})$ distance is 2.14(1) Å. Fig. 2(a) shows a two-dimensional layer containing

$[\text{Mn}(\text{salen})]^+$ units linked by cyano-rhenium clusters, while Fig. 2(b) shows a side view of two layers stacked perpendicular to the (1 0 1) direction. This layer is not a flat sheet, since the angle of the bridging elements $\text{N}(\text{cyano group})-\text{Mn}-\text{N}(\text{cyano group})$ is $169.9(4)^\circ$ and the angles $\text{C}(\text{cyano group})-\text{N}(\text{cyano group})-\text{Mn}$ are $147.4(1)$ and $149.5(1)^\circ$. The unlinked dangling $[\text{Mn}(\text{salen})]^+$ units are directed toward the outside of the layers, and the $\text{C}(\text{cyano group})-\text{N}(\text{cyano group})-\text{Mn}(\text{dangling Mn unit})$ angle is $141.4(1)^\circ$ [Fig. 1(b)]. Two neighboring chains of $-\text{Mn}(\text{salen})-\text{NC}-\text{Re}_6\text{Te}_8-\text{CN}-\text{Mn}(\text{salen})-$ run akin to a twisted rope, such ropes being interconnected by Re_6Te_8 clusters forming two-dimensional sheets. The intermolecular distances between the Mn units *via* the Re cluster are 14.490(9) and 14.38(1) Å, while the distance between the dangling Mn units *via* the Re cluster is 13.788(8) Å. The interlayer distance between the neighboring Mn (dangling Mn unit) atoms is 7.703(2) Å.

This system with cyano-rhenium clusters linked by Mn(salen) complexes is the first complex-bridged cyano-rhenium cluster framework material. Remarkably, this compound shows a unique thick-layered structure which opens the possibility of developing a new class of materials through crystal engineering of appropriate molecular complexes. We are currently investigating other transition metal complexes with high-spin systems which could provide extended network structures with cyano-rhenium clusters.

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- Yield: 39.1%. Anal. Calc. for $\text{C}_{70}\text{H}_{56}\text{Mn}_4\text{N}_{14}\text{O}_8\text{Re}_6\text{Te}_8$: C, 23.49; H, 1.58; N, 5.48. Found: C, 23.48; H, 1.51; N, 5.42%.
- The X-ray diffraction data were collected on a Nonius Kappa-CCD diffractometer using Mo-K α radiation ($\lambda = 0.71071$ Å) at 150 K; crystal size $0.40 \times 0.12 \times 0.06$ mm. The CCD data were integrated and scaled using the DENZO-SMN software package, and the structure was solved and refined using SHELXTL V5.0. All non-hydrogen atoms were located in the calculated positions. *Crystal data*: $\text{C}_{70}\text{H}_{56}\text{Mn}_4\text{N}_{14}\text{O}_8\text{Re}_6\text{Te}_8$, $M = 3579.05$, monoclinic, space group $P2_1/n$ (no. 14) $a = 15.8104(7)$, $b = 13.5732(7)$, $c = 18.9822(5)$ Å, $\beta = 101.738(3)^\circ$, $V = 3988.4(3)$ Å³, $Z = 2$, $\mu(\text{Mo-K}\alpha) = 12.603$ mm⁻¹, 34426 reflections measured, 7847 unique ($R_{\text{int}} = 0.115$) which were used in all calculations, final $R = 0.0552$ ($R_w = 0.1241$) with reflections having intensities greater than 2σ , $\text{GOF}(F^2) = 1.030$. CCDC reference number 152818. See <http://www.rsc.org/suppdata/cc/b1/b104276a/> for crystallographic data in CIF or other electronic format.